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4 Arrangement of Electrons in Atoms

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Modern Chemistry 27 Chapter Test Name Class Date Chapter Test A, continued
____ 4. The electron configuration below violates a. the Pauli exclusion principle. b. the Aufbau principle. c. Hund's rule. d. Both (a) and (c) ____ 5. A photon is emitted from a gaseous atom when an electron moves to its ground state from a(n) a. inner shell. b ...

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Holt Modern Chemistry Review CHAPTER 4: ARRANGEMENT OF ELECTRONS IN ATOMS. The following pages contain the bulk (but not all) of the information for the chapter 4 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test.

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1s 2s 2p 3s 3p 3d 4s 1 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 9. 1.012×10^{12} m What is the wavelength of light that has a frequency of 3.10×10^4 Hz in a vacuum? 10. 3.3×10^{-19} J What is the energy of a photon that has a frequency of 5.0×10^{14} Hz? 32 ARRANGEMENT OF ELECTRONS IN ATOMS MODERN CHEMISTRY

Chapter 4 Modern Chemistry Test

Chapter 4 Test Understand the differences in the energies of different orbital sublevels. Know Aufbau principle, Hund's rule, & Pauli exclusion principle. Identify elements using their electron configuration notation. Determine the electron configuration notation of elements.

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Chapter 4 Modern Chemistry Arrangement Of Electrons In ...

Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

Modern Chemistry Chapter 4 Standardized Test Prep Answers

Holt Modern Chemistry: Chapter 4 Test. Arrangement of Electrons in Atoms. STUDY. PLAY. where can electrons NOT exist. in between orbits. what are atomic energy levels in the atom. the electron orbits. when an atom is in an excited state the electrons are in the _____ orbitals. farther.

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